



**British
Geological Survey**

NATURAL ENVIRONMENT RESEARCH COUNCIL

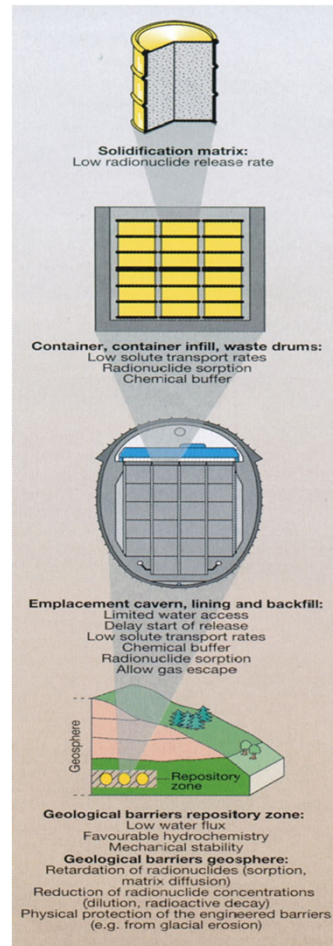
Applied geoscience for our
changing Earth

Harpur Hill High pH Industrial Archaeological Analogue Site: scene setting

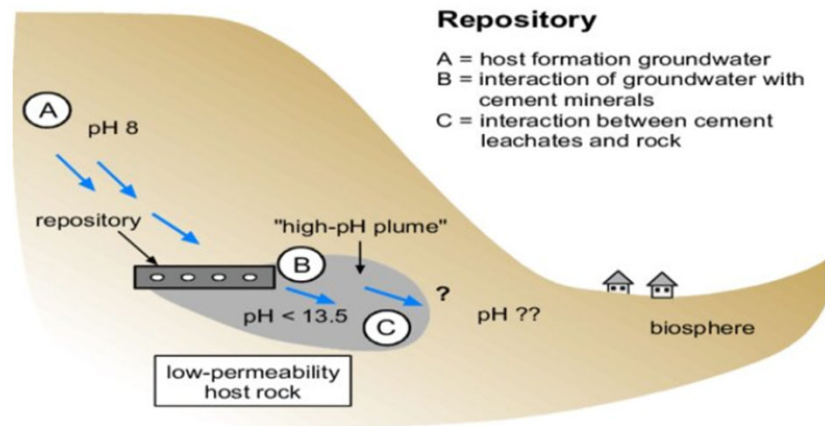
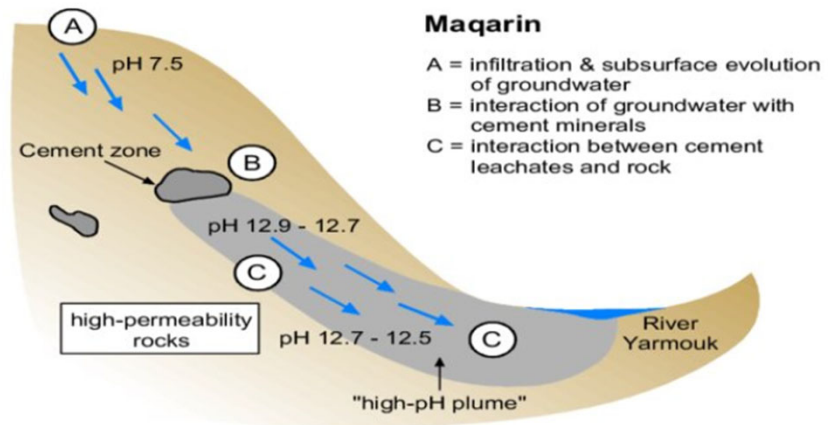
A.E.Milodowski (BGS)

(presented on his behalf by W.R.Alexander)

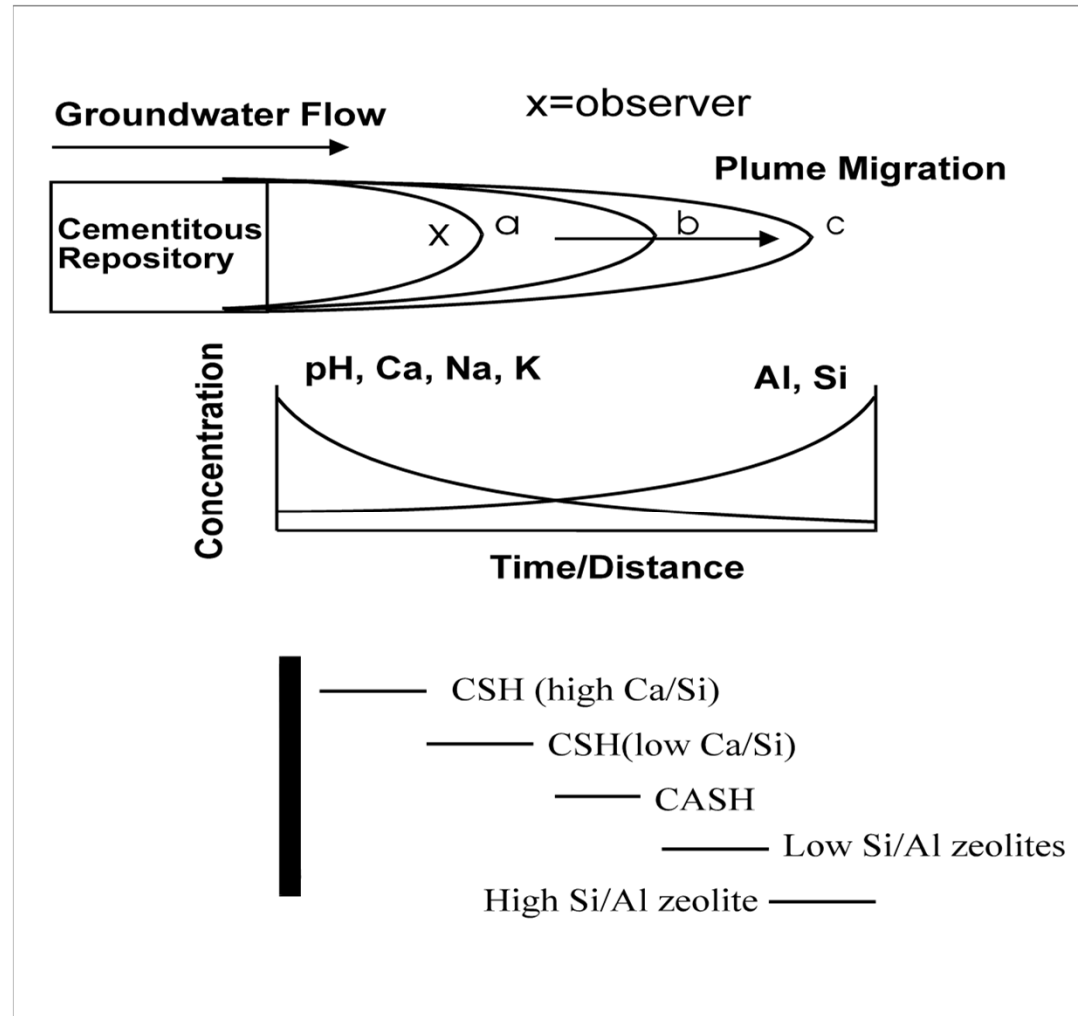
Why Harpur Hill?



Basis of the analogy



Why Harpur Hill?

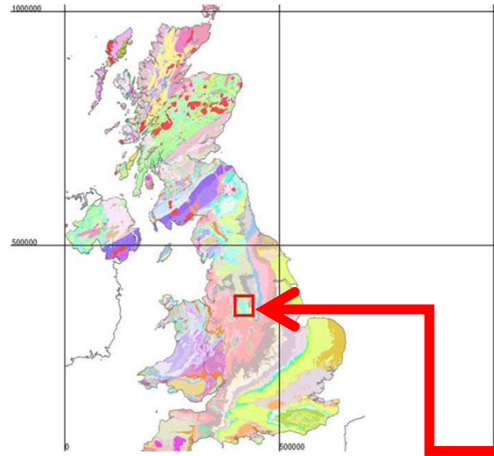


Introduction to Harpur Hill

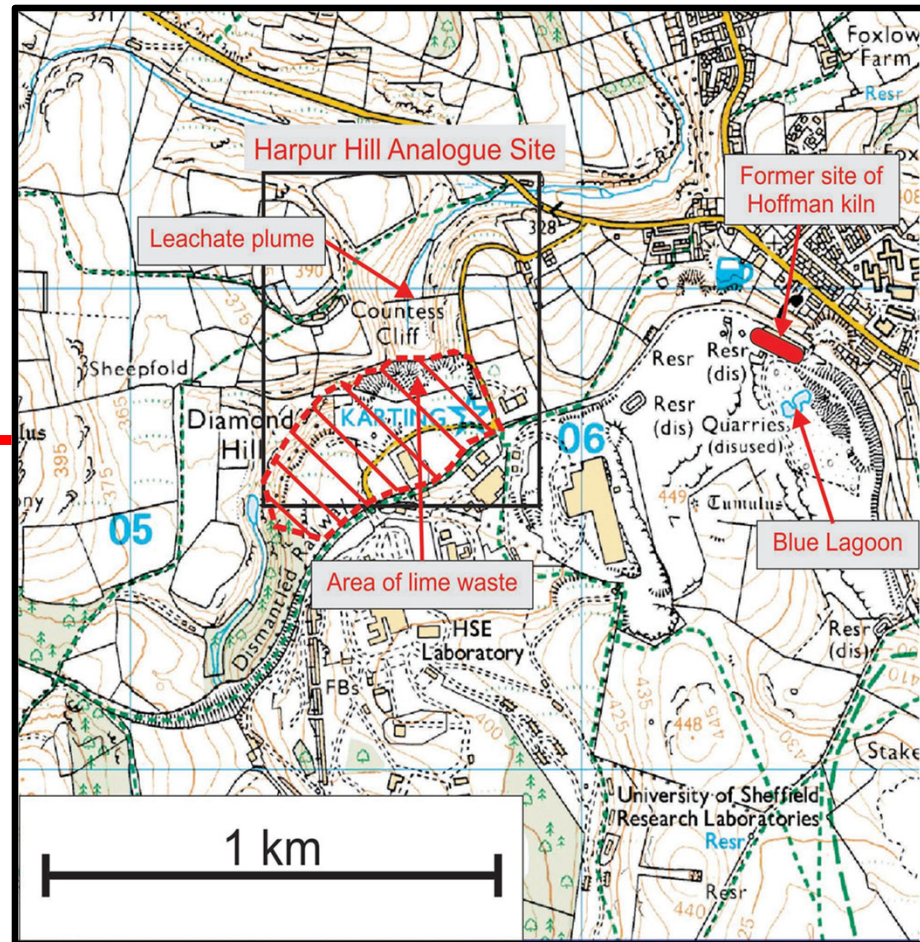


- › Extensive high-pH (pH 12-13) calcium hydroxide [$\text{Ca}(\text{OH})_2$] plume and major associated CaCO_3 tufa deposits
- › Related to shallow groundwater leaching of historic lime kiln clinker waste
- › Waste contains significant residual lime, plus portlandite [$\text{Ca}(\text{OH})_2$] hydration product $\pm$ possible minor calcium silicates.
- › One of similar high pH leachate sites throughout the UK associated with the leaching of lime-bearing and calcium silicate wastes from historic industrial processes (lime kilning, steel and chrome smelting slags)

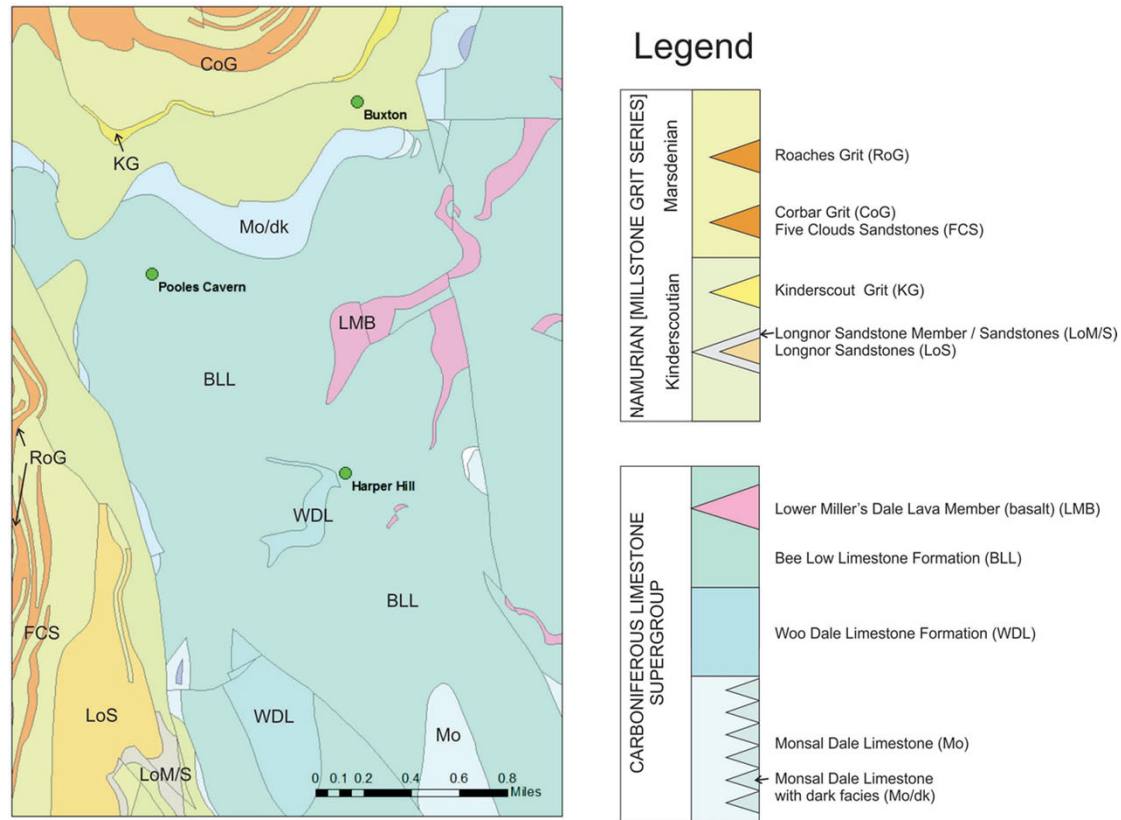
Location and principal site features



Buxton
Peak district, Derbyshire



Geological background



History of lime burning in the Harpur Hill area

Mid 1600's: earliest records of lime kilning on Grin Low

1662: well established lime burning industry in NW area of Grin Low

1734: peak production at Grin Low with over 130 kilns between 17th and 19th centuries

1819: lime kilning has ceased in NW Grin Low and area landscaped (over Poole's Cavern)

1835: First major kilns built at Harpur Hill

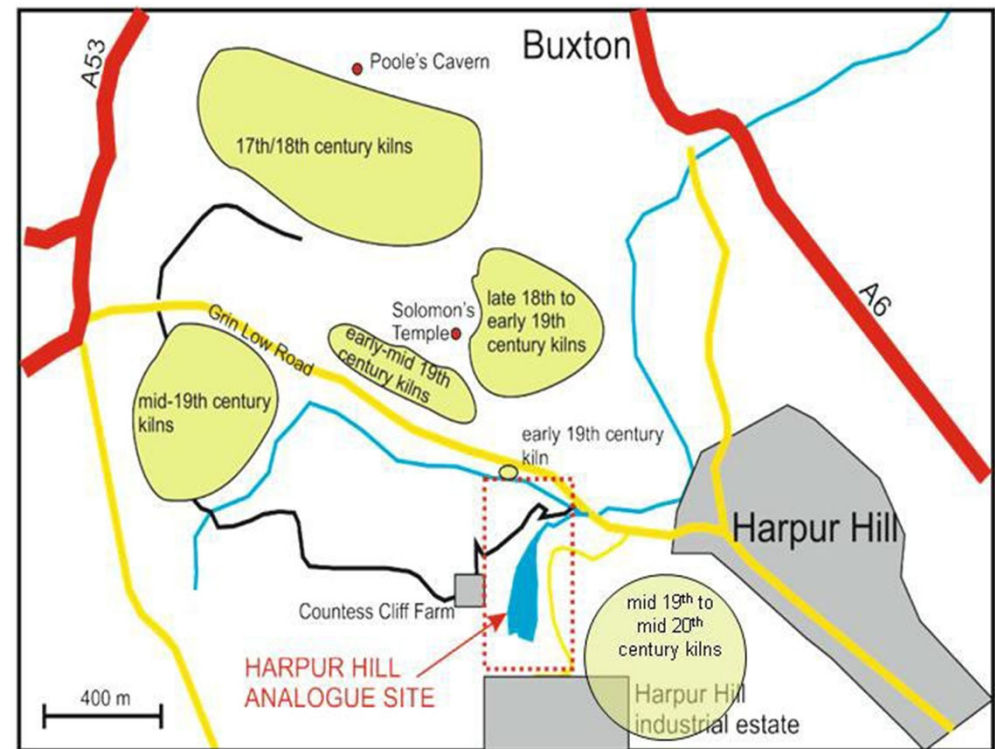
1842: Kilning concentrated to S and SE of Grin Low by this time

1872: Construction of Hoffman kiln at Harpur Hill

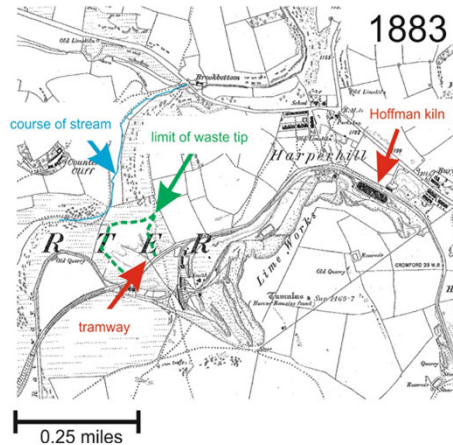
1883-1938: 12 kilns in continuous operation at Harpur Hill

1952: Lime burning ceased at Harpur Hill

1980: Hoffman kiln demolished



Development of lime waste



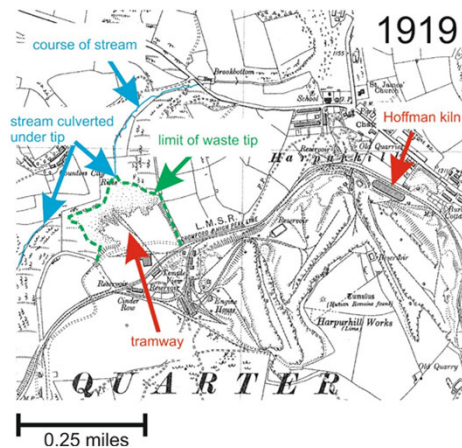
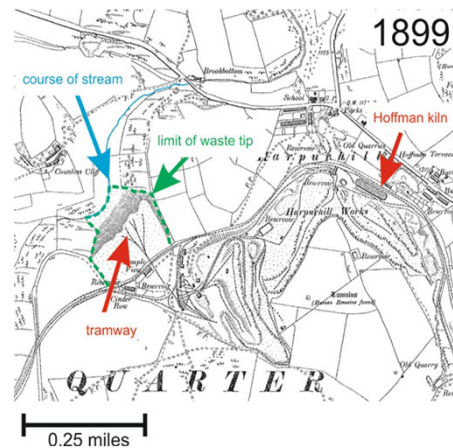
1883: lime waste tip in Brook Bottom valley well-established

1899: lime waste tip reached the stream and is virtually at its maximum extent in the north

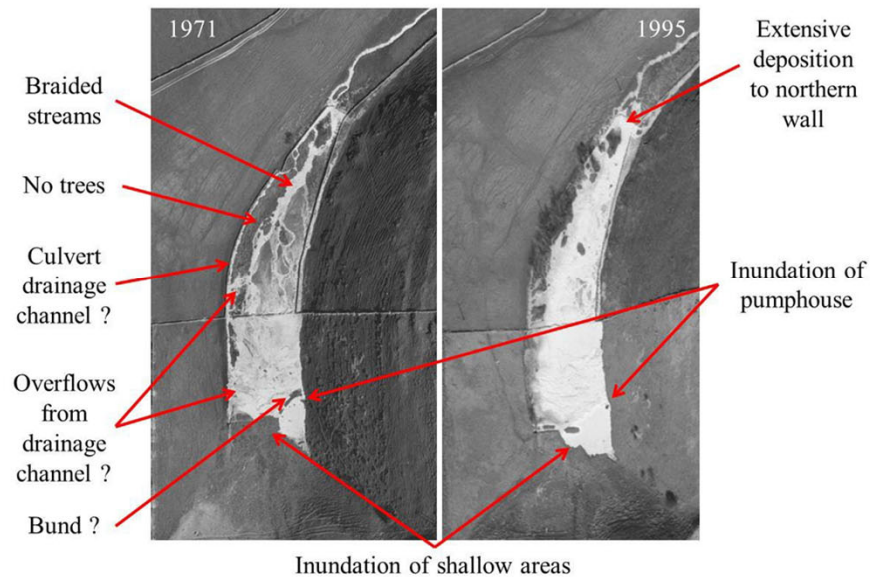
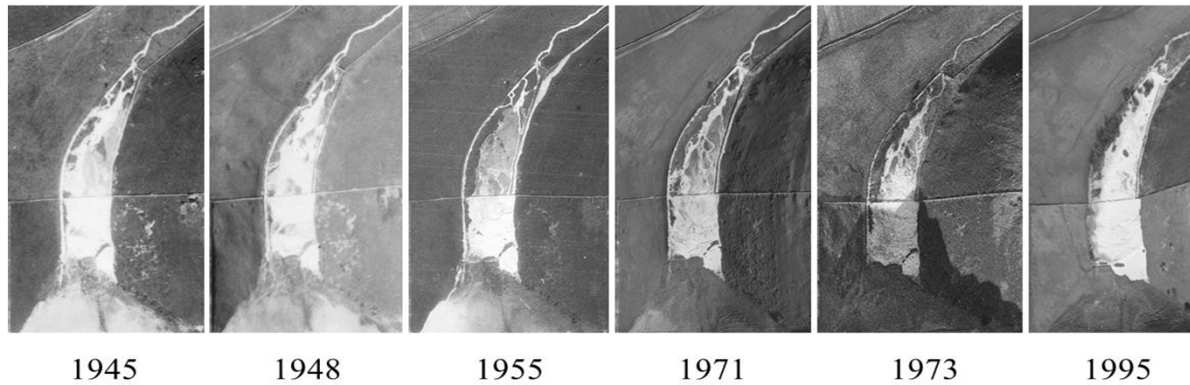
1919: lime waste tip expanded to inundate the stream, further tipping mainly to southwest

Between 1899 and 1919: the stream was culverted (brick culvert in south west of site)

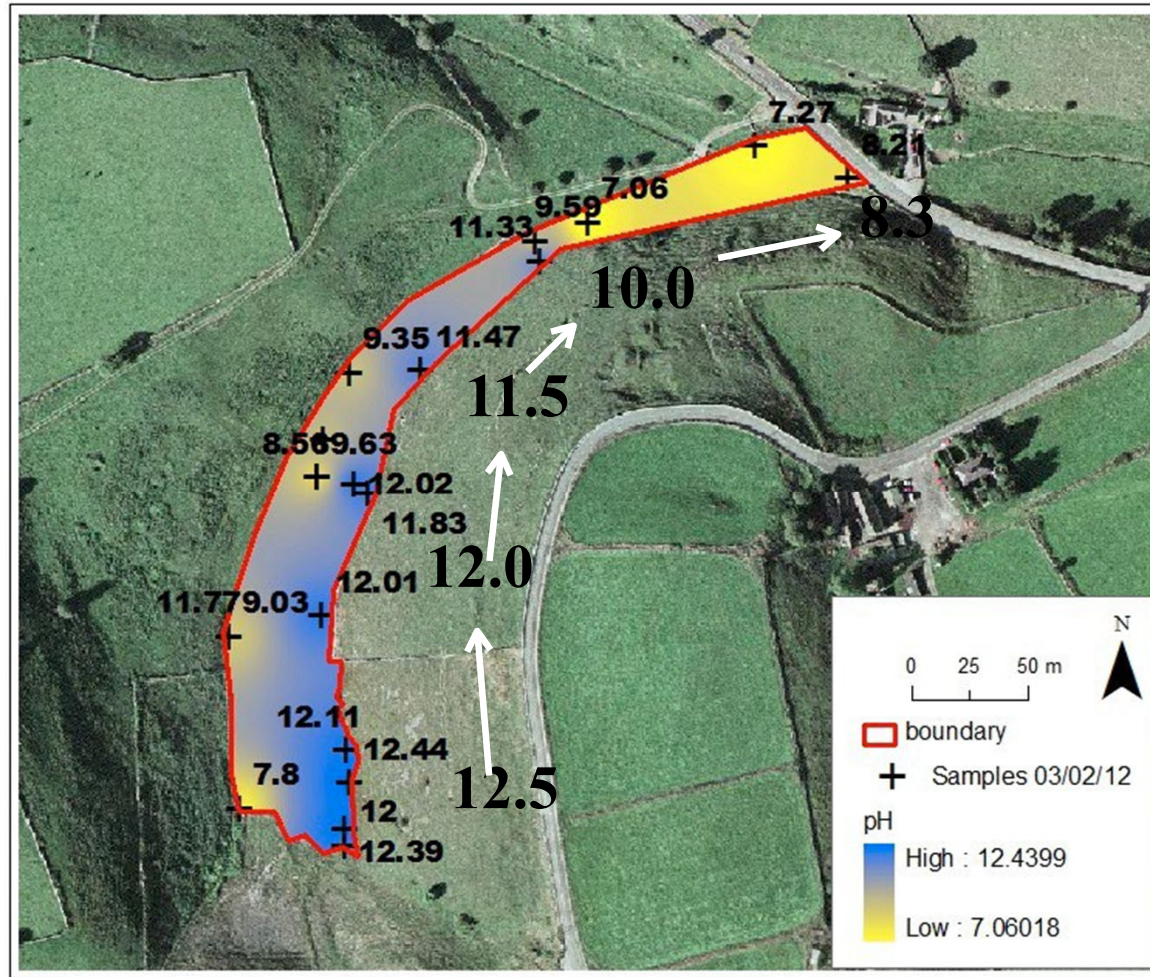
1919: OS maps now show springs ("Rises") at base of the tip, suggesting that leachate plume may have started by now



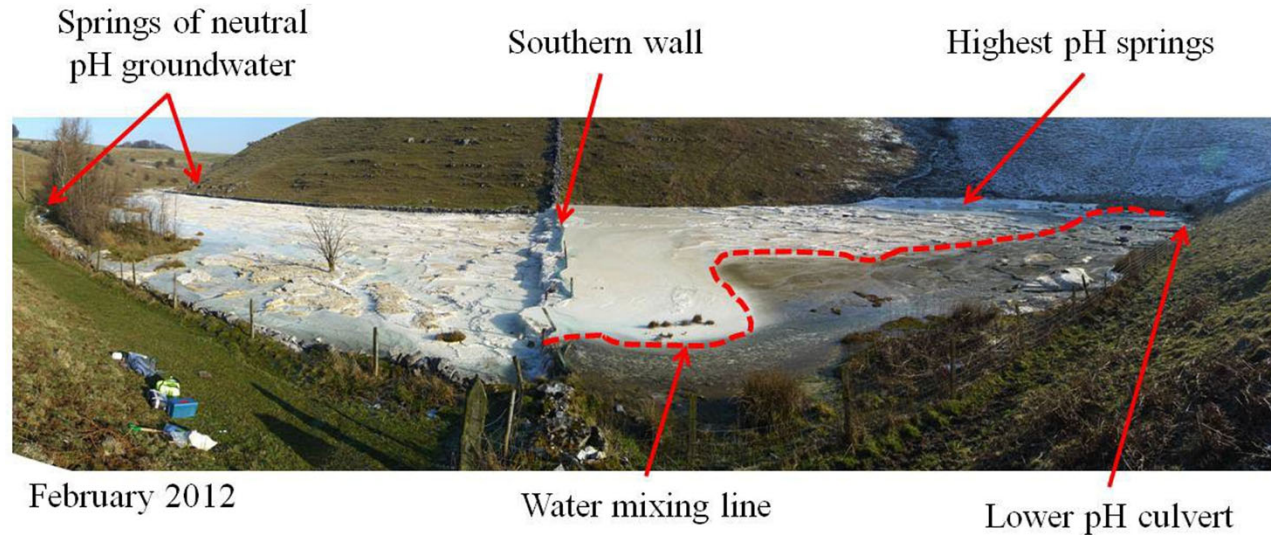
Development of the alkaline plume



pH distribution

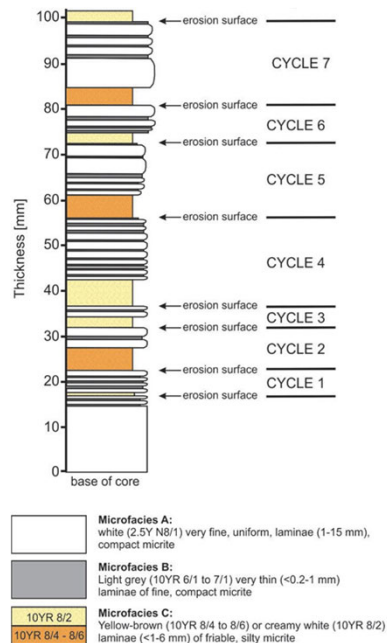


Evolution of the alkaline tufa

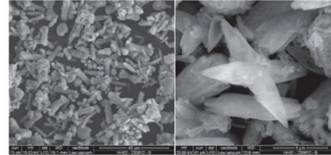


- Complex (bio)geochemistry in surface waters – highly variable, depending on weather conditions (principally rainfall but also temperature)
- Variations on scale of days to weeks
- Dry weather – High pH uniformly across most of the site (~pH11-12.5); stream pH 12.5
- Heavy rainfall / snow melt – pH very variable across the site:
 - Toe of tip / upper bund area pH12.5 (or higher)
 - Stream inflow pH7
- Lower tufa areas – highly variable pH 7 to 12
- Channels with freshwater constantly migrating across the site
- Microbial “blooms” with changing and lower pH – microbiology changing

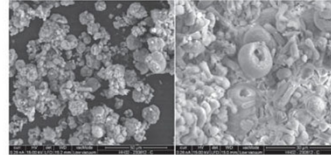
Evolution of the alkaline tufa



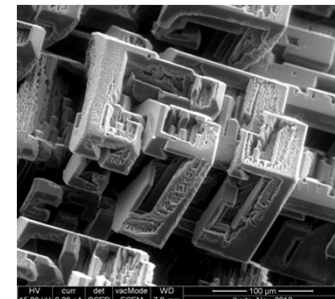
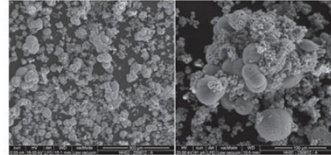
Microfacies A



Microfacies B



Microfacies C

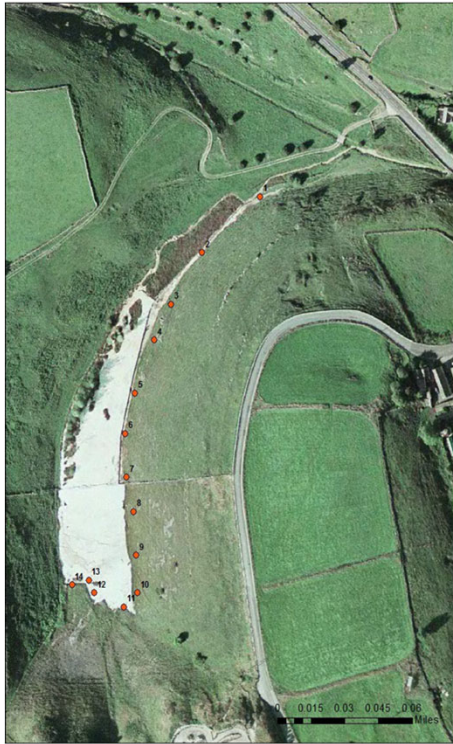


Ikaite
 $\text{CaCO}_3 \cdot 6\text{H}_2\text{O}$

- Cyclic sedimentology reflecting changing environmental (weather, flood events)
- Carbonate precipitation mechanism constantly changing or changing bias:
- Mixing of freshwater and alkaline leachates
- Reaction of atmospheric CO_2 with alkaline leachates
- Ikaite formation in cold weather (<10 C)
- Flooding – erosion, channel flow channels migrating, eroding, reworking and redepositing the calcareous sediments
- Organic blooms in wet weather (lower pH – higher biological activity)

Inventory assessment

Detailed walkover survey



Inventory summary

Plant species (living and dead) including (High interest)	Metals and metallic materials (High interest)	Building materials (Moderately low interest)	Naturally derived materials (High interest)	Plastics (Low interest)	Other (Moderately high interest)
Grass (cellulosic)	Copper metal	Bricks	Wood	Polythene	Newspaper
Trees	Lead metal	Concrete	Sandstone	Polystyrene	Glass
Herbaceous plants, horsetails, algae and mosses	Gold metal	Bitumen	Soils, inc. clays , silicates etc	PET	Various kiln glazes
	Aluminium alloy		Limestone		
	Various steels inc. galvanised steel		Slag (lime and silicate)		
	Brass		Rubber		
	Plastic coated steel and copper metal		Part burnt coke and coal		

Relevance and issues

- Many of the materials are in the oxic, near-surface waters or exposed to air: {How relevant is this to reducing repository environment?}
- Near surface waters constantly oscillating in chemistry and microbiology {not anticipated in the repository environment}
- Focus should be on materials contained deeper with the sedimentary pile, where conditions more stable regardless of weather and where it may be reducing {more relevant to repository environment} – but locating these may be problematic
- Cellulosic materials present and degrading {grass remains present at original soil- sediment interface and wood (trees, posts) buried deep in the sedimentary pile}
- Alkaline interaction with clays and other silicate minerals at depth beneath the alkaline tufa deposits. Appear to be associated with some microbial mineralisation features (Fe alteration)
- Stable high pH (>12.5) in “upper banded area” at depth (>0.5 m) regardless of weather conditions {most relevant to the repository environment}
- Formation of metastable ikaite related to carbonation reactions with Ca(OH)_2 porefluids may be important with regard to cement alteration through carbonation, under permafrost and cold environmental conditions

Some images of the site



Some images of the site



Some images of the site



Last word – the Blue Lagoon



Bibliography

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- Ongoing microbiology studies addressing the 2023 NWS microbiology roadmap by Jon Lloyd's group at the University of Manchester (e.g. Rebecca Snow's previous presentation) and BGS